

Check Your Result at: <u>www.myrankers.com</u> Reward Ceremony Date: <u>16th Feb 2025</u>

Student's Name: -

Pg.(2)

Physics (Section – A)

- 1. A block of mass 1kg connected with a spring of spring constant 4 N/m lies on a smooth horizontal surface. The block is compressed by 20 cm and then released. Find the minimum time after which the elongation in spring becomes 10 cm.
- (A) π sec (B) $\frac{\pi}{2}$ sec (C) $\frac{\pi}{3}$ sec (D) $\frac{\pi}{4}$ sec 2. All resistance in diagram are in ohms. Find the effective resistance between the points Q and S. (A) 2 Ω
 - (B) 3 Ω
 - (C) 4 Ω
 - (D) 6 Ω

 $P \xrightarrow{A}_{3} \xrightarrow{R}_{4} \xrightarrow{R}_{5} \xrightarrow{R}_{6} \xrightarrow{R}_{7} \xrightarrow{R}_{7}$

3. A charge +q is fixed at each of the points $x = x_0$, $x = 3x_0$, $x = 5x_0$, upto infinity on the x-axis and a charge -q is fixed at each of the points $x = 2x_0$, $x = 4x_0$, $x = 6x_0$, upto infinity. Here x_0 is a positive $\frac{Q}{4\pi \in_0 r}$ constant. Take the electric potential at a point due to a charge Q at a distance

r from it to be. Then the potential at the origin due to the above system of charges is:

- (A) $\frac{q}{4\pi\epsilon_0 x_0\ell n2}$ (B) $\frac{q}{8\pi\epsilon_0 x_0\ell n2}$ (C) $\frac{q\ell n2}{8\pi\epsilon_0 x_0}$ (D) $\frac{q\ell n2}{4\pi\epsilon_0 x_0}$
- 4. The uniform rod AB of mass m is released from rest when $\beta = 60^{\circ}$. Assuming that the friction force between end A and the surface is large enough to prevent sliding.

What is the initial angular acceleration of the rod?



Pg.(3)

5. In figure the mass of body A is four times as great as that of body B. The system is released from rest find acceleration of block B. $(g = 10 \text{ m/s}^2)$ (A) 2.5 m/s²



6. The magnetic field at the origin due to a current element i $\vec{d\ell}$ placed at a position \vec{r} is

(A)
$$\frac{\mu_0 i}{4\pi} \frac{\vec{d\ell} \times \vec{r}}{r^3}$$
 (B) $-\frac{\mu_0 i}{4\pi} \frac{\vec{r} \times \vec{d\ell}}{r^3}$ (C) $\frac{\mu_0 i}{4\pi} \frac{\vec{r} \cdot \vec{d\ell}}{r^3}$ (D) $-\frac{\mu_0 i}{4\pi} \frac{\vec{d\ell} \times \vec{r}}{r^3}$

7. An inductor $\left(L = \frac{1}{100\pi}H\right)$, a capacitor $\left(C = \frac{1}{500\pi}F\right)$ and a resistance (3 Ω) is connected in series with an AC voltage source as shown in the figure. The voltage of the AC source is given as V = 10 cos(100 π t) volt. What will be the potential difference between A and B?

(A) 8 cos(100
$$\pi$$
t – 127°) volt

- (B) 8 cos(100 π t 53°) volt
- (C) $8 \cos(100 \pi t 37^{\circ})$ volt
- (D) 8 cos(100 π t + 37°) volt



----- Rough Work -----

8. Two blocks of mass 2kg and 3kg are connected to a spring of force constant 120 N/m. If both the blocks are given a velocity 1.0 m/s each directed away from each other then find the maximum elongation in spring.

(A) 20cm (B) 10cm (C) 5cm (D) 15cm

9. A curved mirror is kept such that it's cross section lies in x - y plane whose equation is given as $y = \frac{2L}{\pi} \sin\left(\frac{\pi x}{L}\right)$. Find the point at which a light ray travelling parallel to x-axis must incident in order to get reflected parallel to y-axis. \uparrow (y)



- 10. A hollow hemisphere rests on a sufficiently rough surface inclined at an angle θ with horizontal, with it's base being vertical. For what value of θ the sphere will remain in equilibrium. (A) 15°
 - (B) 30°
 - (C) 45°
 - (D) 60°



----- Rough Work -----

Chemistry (Section – B)

- Which of the following imparts green colour to the burner flame? 11. (B) Na(OMe) (A) $B(OMe)_3$ (C) $Al(OPr)_3$ (D) $Sn(OH)_2$
- 100g of an ideal gas is kept in a cylinder of 416 L volume at 27° C under 1.5 bar pressure. The 12. molar mass of the gas is g mol⁻¹. (Given: $R = 0.083 \text{ L bar } \text{K}^{-1} \text{ mol}^{-1}$) (B) 4 (C) 5 (A) 6 (D) 7
- 13. Which of the following statements is not correct? (A) Silicon is extensively used as a semiconductor (B) Carborundum is SiC (C) Silicon occurs in free state in nature
 - (D) Mica contains the element silicon
- In the mixture of (Na₂CO₃ + NaHCO₃) volume of HCl required is x mL with phenolphthalein 14. indicator and y mL with methyl orange indicator in the same titration. Hence volume of HCl for complete reaction of NaHCO₃ is:
 - $(C)\frac{x}{2}$ (D) (y - x)(A) 2x(B) y
- 15. Give product for following reaction.



----- Rough Work -----

ΩН

16.
 4.5 moles each of hydrogen and iodine is heated in a sealed ten litre vessel. At equilibrium, 3 moles of HI were found. The equilibrium constant for H₂(g) + I₂(g)
$$\approx$$
 2HI_(g) is

 (A) 4
 (B) 3
 (C) 2
 (D) 1

 17.
 Which of the following is a non-reducing sugar?

 (A) Lactose
 (B) Fructose
 (C) Sucrose
 (D) Maltose

 18.
 The most basic compound among the following is

 (A) Acetanilide
 (B) Benzylamine
 (C) *p*-Nitro aniline
 (D) Aniline

 19.
 Calculate the volume of water (in mL) required to dissolve 0.1 g lead (II) chloride to get a saturated solution (K_{sp}of PbCl₂ = 3.21 × 10⁻⁸, atomic mass of Pb = 207 u). Report your answer by rounding it up to nearest whole number.

 (A) 160
 (B) 170
 (C) 190
 (D) 180

 20.
 Ph-CH-CH-COOH
 (i) Ag₂O

 vrounding it up to nearest whole number.
 (A) 160
 (B) 170

 (A) Ph-CH-CH-CH₂-Br
 (B) Ph-C-CH-CH₃
CH₃ CH₃
(C) Ph-CH-CH-Br
 (D) Ph-CH-CH₂-Br
 (C) Ph-CH-CH-Br
 (D) Ph-CH-CH₂-Br

Pg.(7)

Math (Section – C)

21. $\lim_{x \to 0} \frac{x \tan 2x - 2x \tan x}{(1 - \cos 2x)^2} \text{ equals}$ (A) $-\frac{1}{2}$ (B) $\frac{1}{4}$ (C) $\frac{1}{2}$ (D) 1

22. A bag contains 4 red and 6 black balls. A ball is drawn at random from the bag, its colour is observed and this ball along with two additional balls of the same colour are returned to the bag. If now a ball is drawn at random from the bag, then the probability that this drawn ball is red, is $(A)\frac{3}{4} \qquad (B)\frac{3}{10} \qquad (C)\frac{2}{5} \qquad (D)\frac{1}{5}$

- 23. If $f(x) = \frac{(\tan 1^\circ)x + \log_e(123)}{x \log_e(1234) (\tan 1^\circ)}$, x > 0, then the least value of $f(f(x)) + f(f(\frac{4}{x}))$ is (A) 4 (B) 2 (C) 0 (D) 8
- 24. Suppose A is any 3×3 non-singular matrix and (A 3I)(A 5I) = 0, where $I = I_3$ and $0 = O_3$. If $\alpha A + \beta A^{-1} = 4I$, then $\alpha + \beta$ is equal to (A) 13 (B) 7 (C) 12 (D) 8
- 25. The value of $\int_{1/e}^{\tan x} \frac{tdt}{1+t^2} + \int_{1/e}^{\cot x} \frac{dt}{t(1+t^2)}$ is equal to (A) -1 (B) 1 (C) 0 (D) e

----- Rough Work -----

26.	The value of cosec 20° (A) 0	tan 60° – sec 20° is (B) 1	(C) 2	(D) 4
27.	The sum of all the solution $(A) 1 + \log_6(8)$	tions of the equation ((B) log ₈ (6)	8) ^{2x} - 16 · (8) ^x + 48 (C) 1 + $\log_8(6)$	= 0 is: (D) $\log_8(4)$
28.	If x satisfies the inequal (A) $\left(\frac{1}{5}, 5\right)$	ality $\log_{25} x^2 + (\log_5 x)$ (B) $\left(\frac{1}{25}, 5\right)$	$(C) \left(\frac{1}{5}, 25\right)^2$	to (D) $\left(\frac{1}{25}, 25\right)$
29.	Let $a_1, a_2, a_3,, a_n,$ is equal to (A) 306	be in A.P. If $a_3 + a_7 =$ (B) 204	+ a_{11} + a_{15} = 72, then (C) 153	n the sum of its first 17 terms (D) 612
30.	Number of common ta (A) 1	ngents of $y = x^2$ and y (B) 2	$y = -x^2 + 4x - 4$ is (C) 3	(D) 4

----- Rough Work -----

Pg.(8)